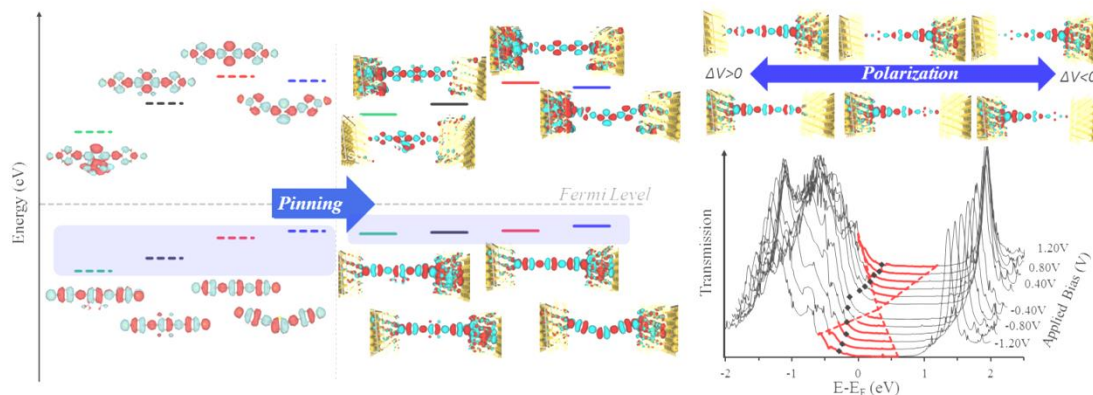




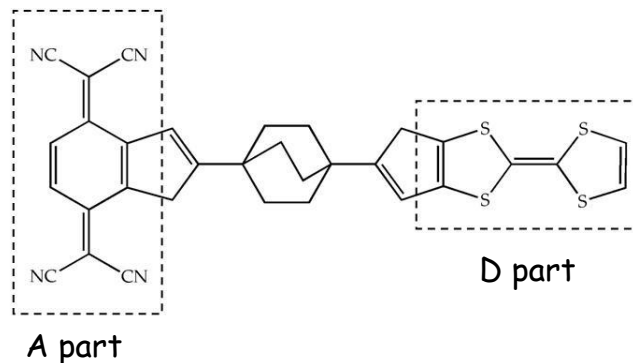
Fermi Level Pinning and Orbital Polarization Effects in Molecular Junctions

Sandra Rodríguez González, Jérôme Cornil, Colin Van Dyck

Sandra.RODRIGUEZGONZALEZ@umons.ac.be

Laboratory for Chemistry of Novel Materials

1974 : Aviram and Ratner



Can we develop all the basic functions of common electronic components at nanoscale



Angewandte
Communications

VIP Self-Assembled Monolayers

DOI: 10.1002/anie.201201448

The Rate of Charge Tunneling through Self-Assembled Monolayers Is Insensitive to Many Functional Group Substitutions**

Hyo Jae Yoon, Nathan D. Shapiro, Kyeng Min Park, Martin M. Thuo, Siowling Soh, and George M. Whitesides*

Chem Soc Rev

RSC Publishing

TUTORIAL REVIEW

View Article Online
View Journal | View Issue

Molecule–electrode interfaces in molecular electronic devices

Chuancheng Jia^a and Xuefeng Guo^{*ab}

Cite this: *Chem. Soc. Rev.*, 2013, 42, 5642

Nanoscale
Horizons



COMMUNICATION

View Article Online
View Journal | View Issue



Cite this: *Nanoscale Horiz.*, 2016, 1, 399

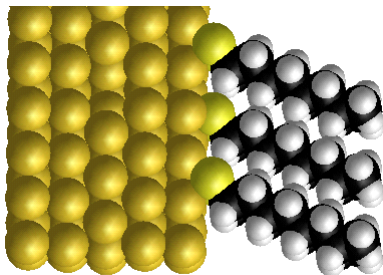
Received 12th May 2016,
Accepted 31st May 2016

DOI: 10.1039/c6nh00088f

Dipole effects on the formation of molecular junctions†

Sachie Tanimoto, Makusu Tsutsui,* Kazumichi Yokota and Masateru Taniguchi

Electronic processes in a SAM/metal interfaces

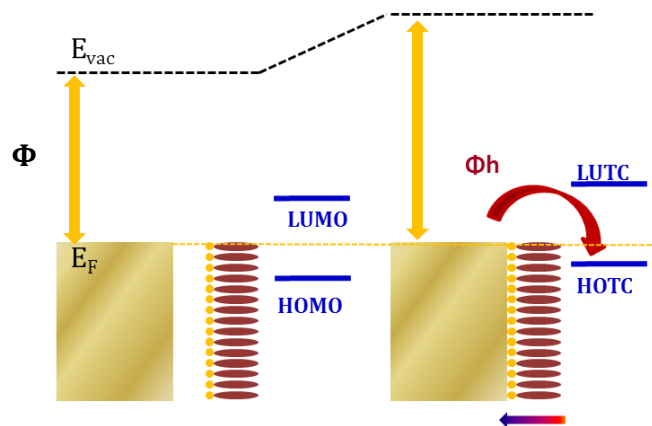


$$\Delta\Phi = \Delta V_{\text{SAM}} + \text{BD}$$

$\Delta\Phi$ = Total Work Function Shift.

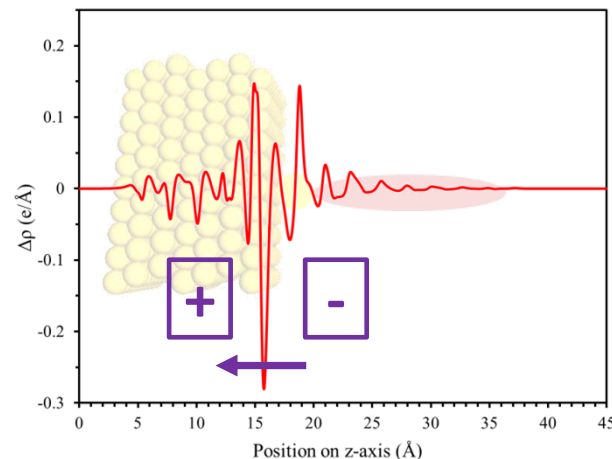
BD = Bond Dipole, interfacial electronic reorganization upon formation of the Au-S bond and surface reconstruction effects.

ΔV_{SAM} = molecular contribution from the dipole supported by the molecular backbone.

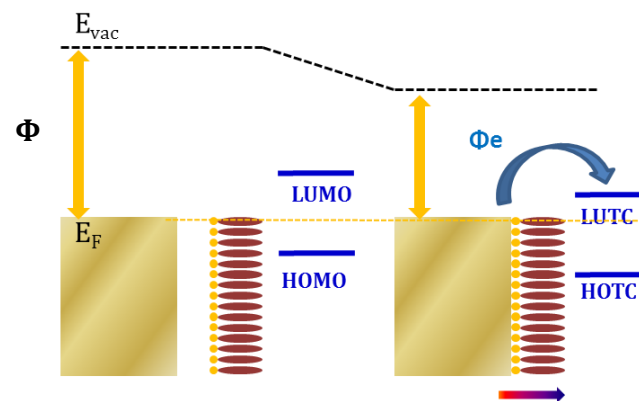


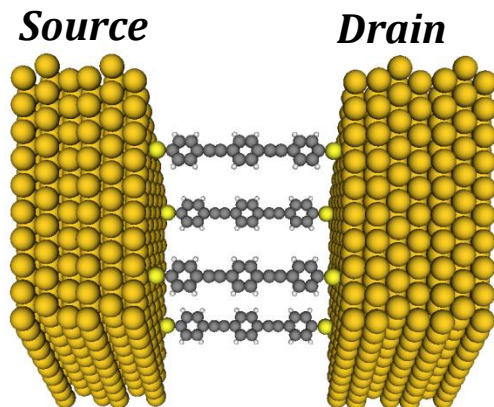
Electron Density Reorganization

$$\Delta\rho(z) = \rho_{\text{Au}(111)/\text{mol}}(z) - (\rho_{\text{mol}}(z) + \rho_{\text{Au}(111)}(z))$$

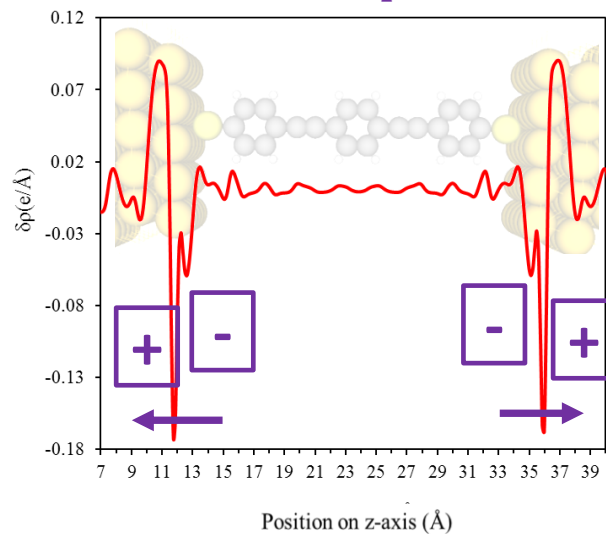


Interface Dipole (BD)

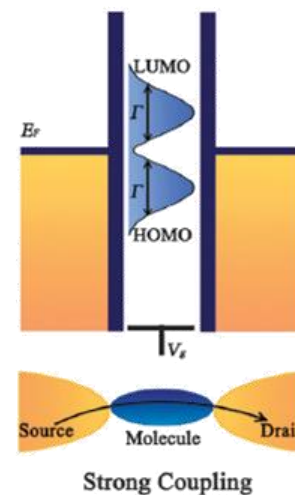
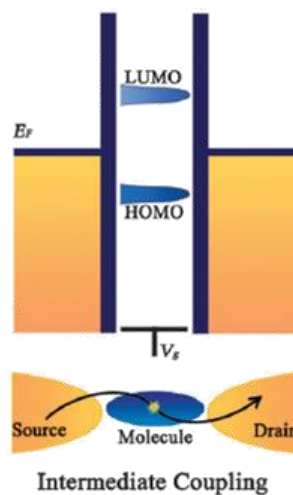
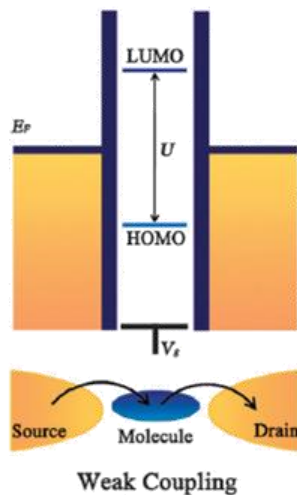




Interface Dipoles



Voltage = 0V



Broadening

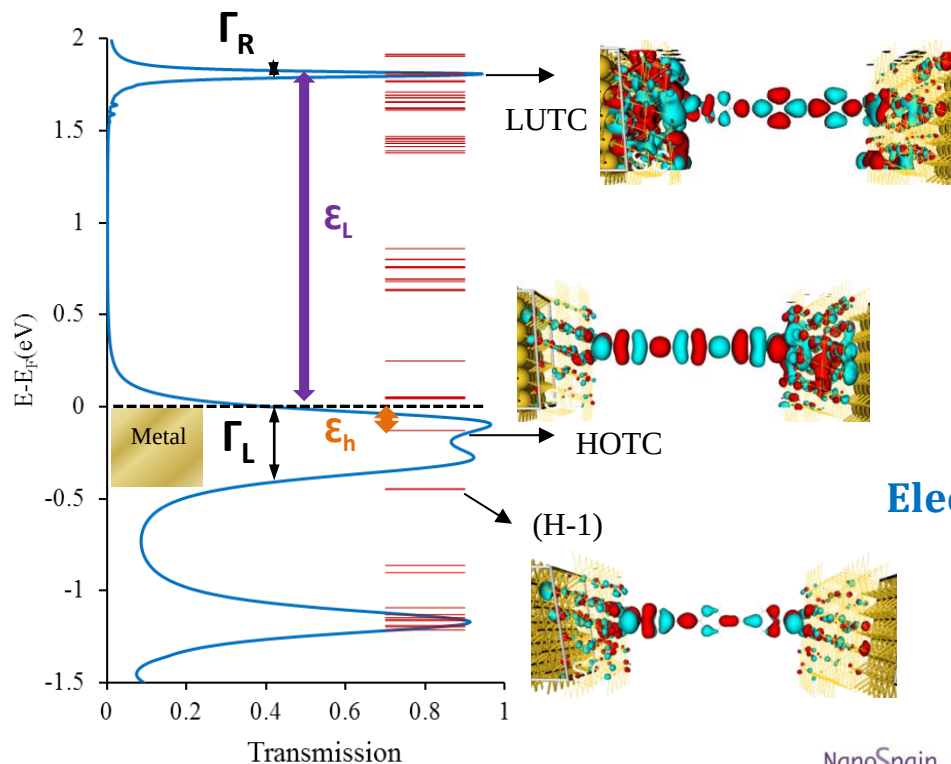
Energy shift

- Energy alignment of the molecular TC relative to the metal Fermi level: ϵ_h/ϵ_L
- Interplay between interfacial and intramolecular electronic couplings.

* **Voltage = 0V**

Transmission Spectrum: gives the probability of one electron across the junction through the molecule in a coherent regime.

(MPSH) analysis



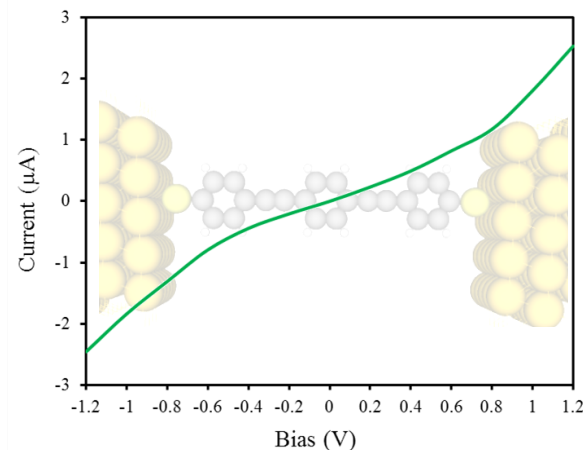
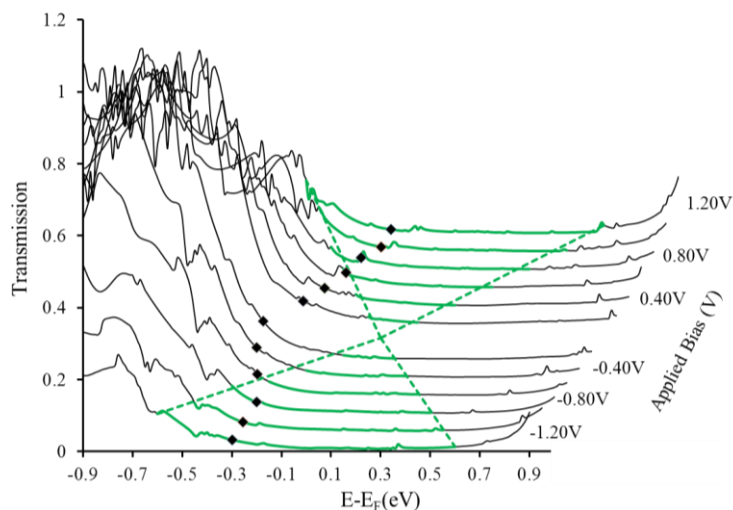
Interfacial coupling (Γ_L/Γ_R) → hybridization:
Strong hybridization ↔ Wide peak

Electronic Delocalization ↔ Intensity of the peak



* **Voltage $\neq 0V$**

Transmission Spectrum within a transmission window

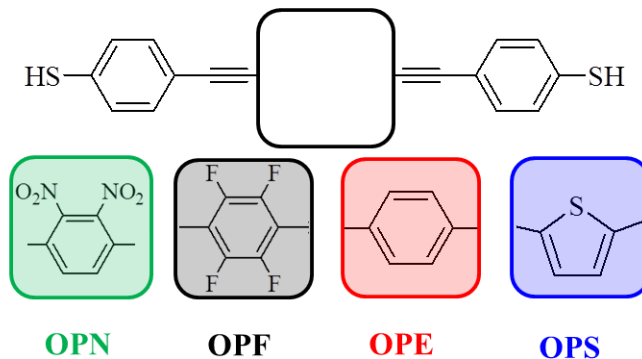


$$I(V) = \frac{2e}{h} \int_{\mu_l}^{\mu_r} T(E, V) [f(E - \mu_l) - f(E - \mu_r)] dE$$

Landauer-Büttiker Formalism

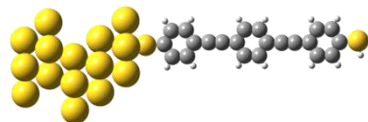
- $T(E, V)$: transmission probability at applied bias (V)
- $\mu_{l,r} = E_F \pm eV/2$: chemical potential of the electrodes.
- E_F : metal Fermi energy
- $f(E - \mu)$: Fermi Dirac distribution function.

Molecular Wires

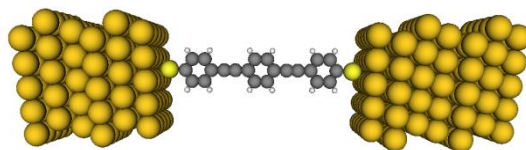


Objectives

1. Modulation of the Work Function of a metal substrate.

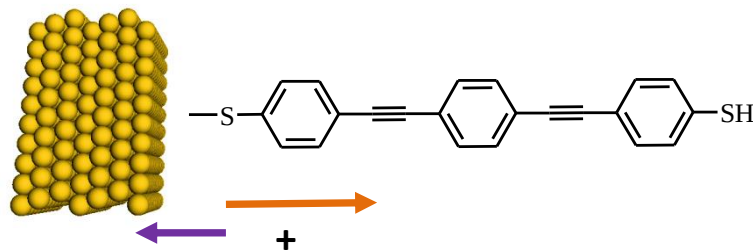


2. Energetic alignment of the molecular transport channels relative to the metallic Fermi level.
3. Conductivity , simulation the I-V curves and analyzing them from the evolution of the transmission spectra under bias.



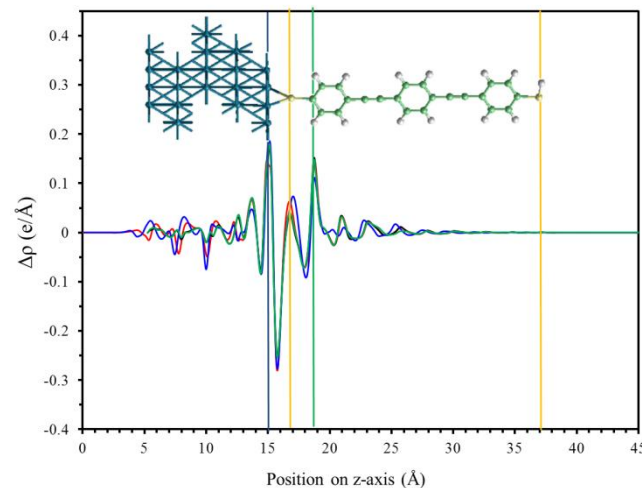
- DFT
- GGA/PBE
- Troullier-Martin Pseudopotentials
- PBC
- SIESTA

- DFT-NEGF
- GGA/revPBE
- Troullier-Martin Pseudopotentials
- PBC
- ATK



$$\Delta\Phi = \Delta V_{\text{SAM}} + \text{BD}$$

Electron Density Reorganization

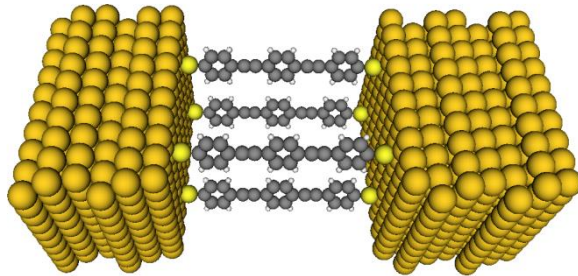


eV	$\Delta\Phi_{\text{theo}}$	Φ_{theo}	ΔV_{SAM}	BD	$\Delta\Phi_{\text{exp}}$	Φ_{exp}
OPN	-0.969	3.862 (0.001)	-2.160	1.191	*	*
OPF	-1.051	3.780 (0.081)	-2.293	1.242	-1.025	4.175 (0.005)
OPE	-0.970	3.861 (0.000)	-2.306	1.336	-1.020	4.180 (0.000)
OPS	-1.071	3.760 (0.101)	-2.262	1.191	-0.926	4.274 (0.094)

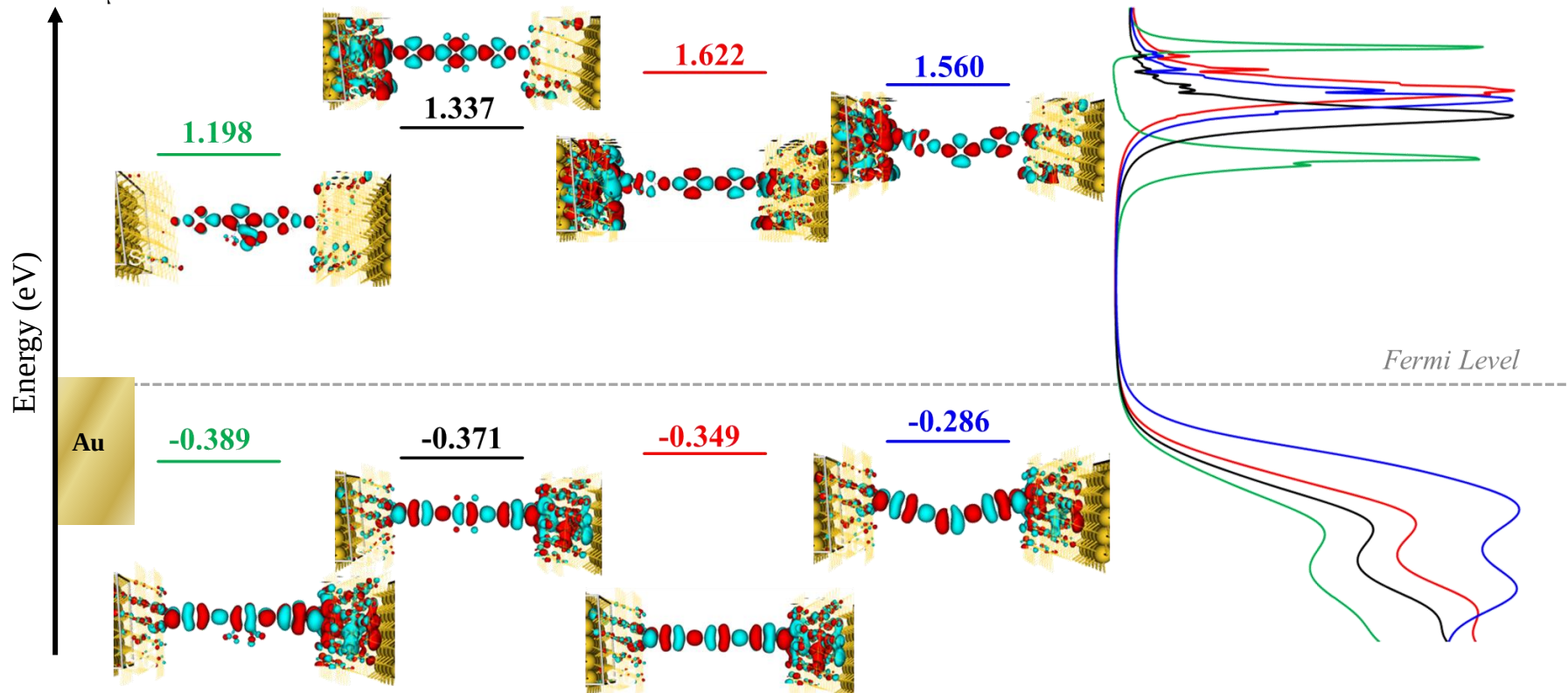
↓Φ upon absorption with respect to the clean Au (111) surface

Substitution weakly modifies the Φ of Au (111) ★

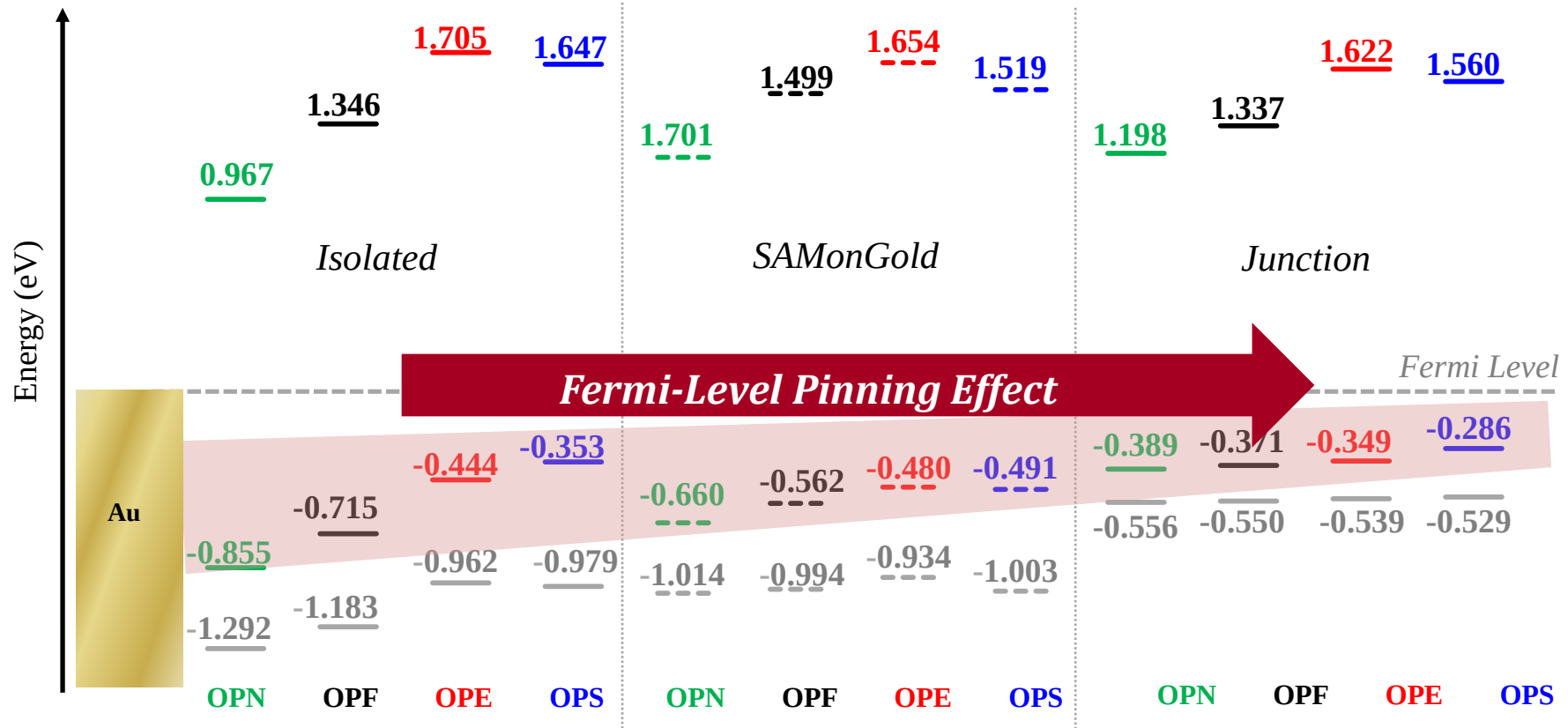
Transmission and MPSH analysis at equilibrium



	T 0 V	L(Å)	ϵ_h (eV)
OPN	0.016	20.29	0.389
OPF	0.019	20.29	0.371
OPE	0.021	20.29	0.349
OPS	0.039	19.30	0.286



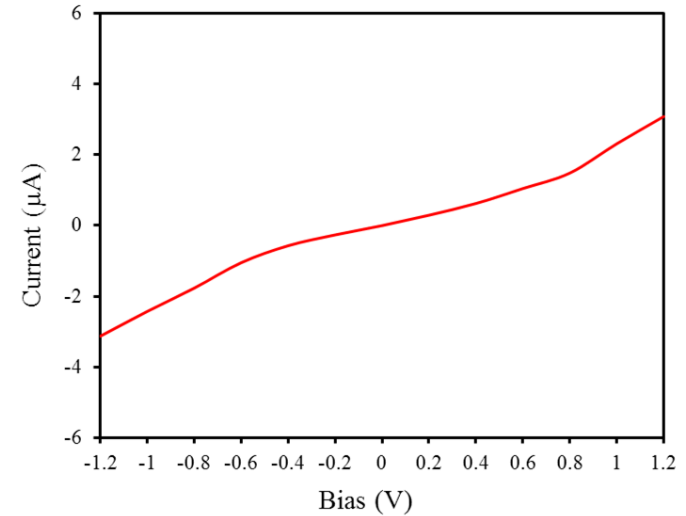
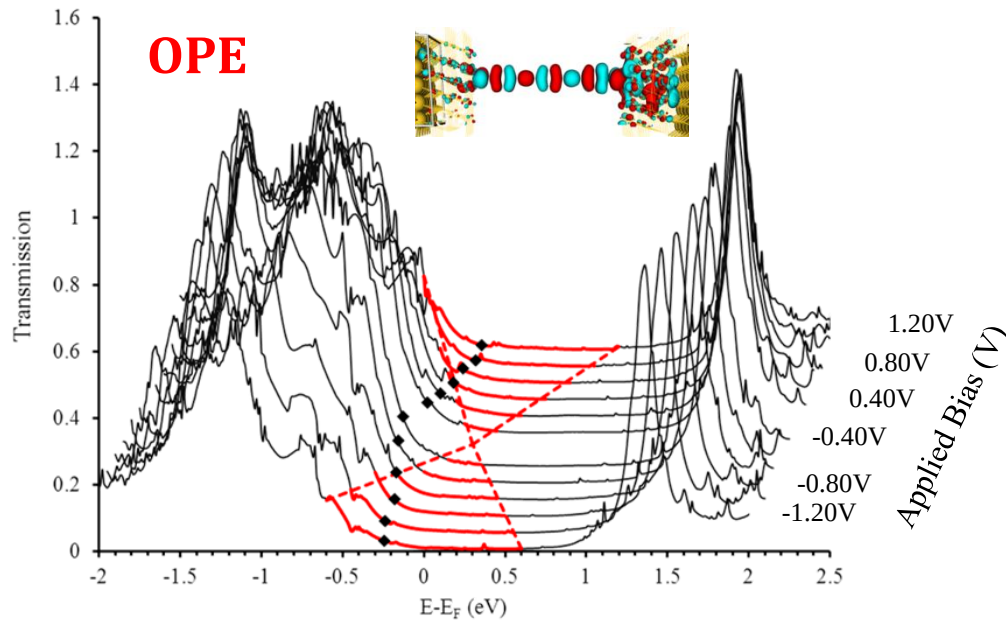
Energy-Level Alignment at equilibrium



ϵ_h (eV)	Iso	SAM on Gold	Junction	Exp
OPN	0.855	0.660	0.389	*
OPF	0.751	0.562	0.371	0.61
OPE	0.444	0.480	0.349	0.59
OPS	0.353	0.491	0.286	0.51

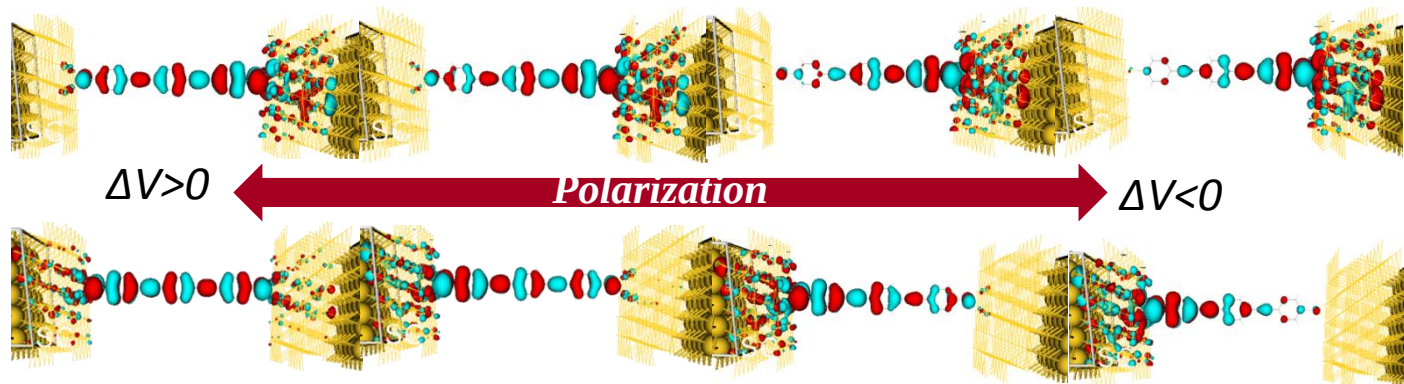
* CP-AFM measurements (Not for **OPN**, yet)

Application of Bias

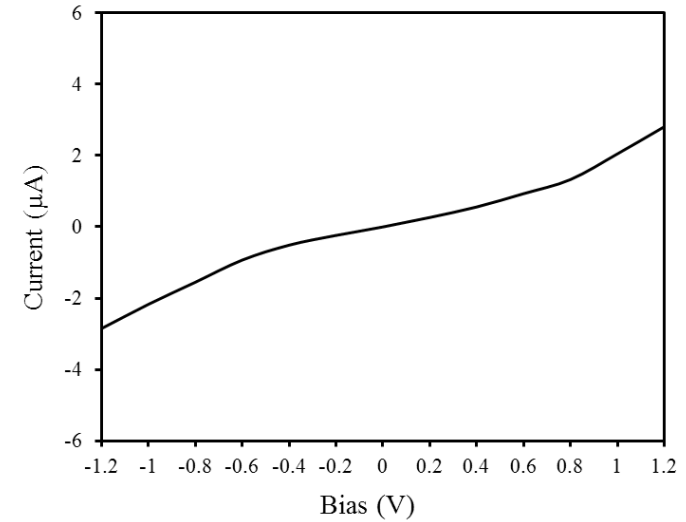
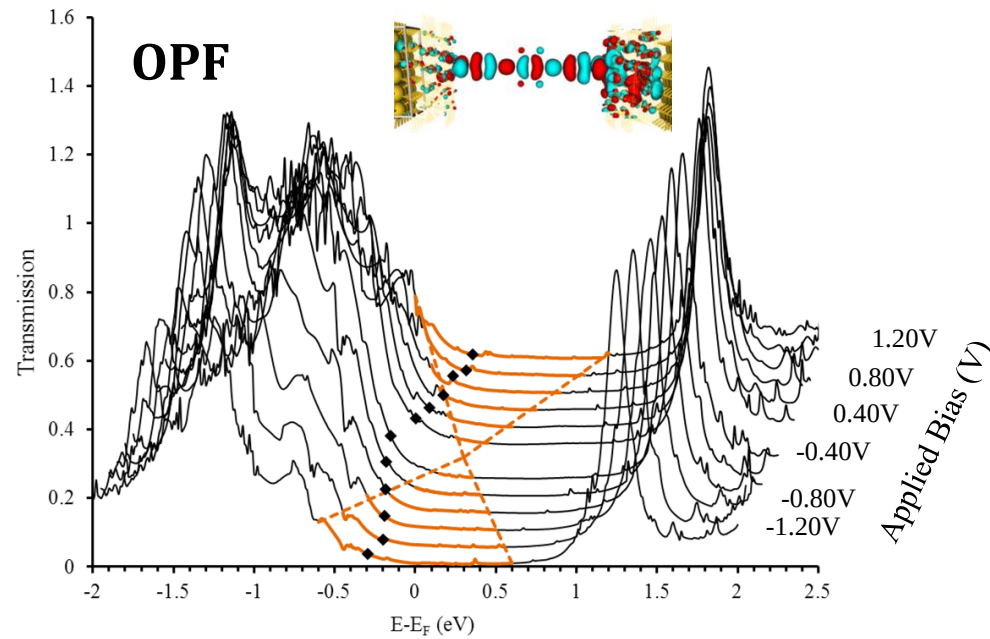


CONTINUOUS INCREASE OF CURRENT

HOMO follows the destabilized electrode! $\rightarrow$ *Intramolecular > Interfacial Coupling*

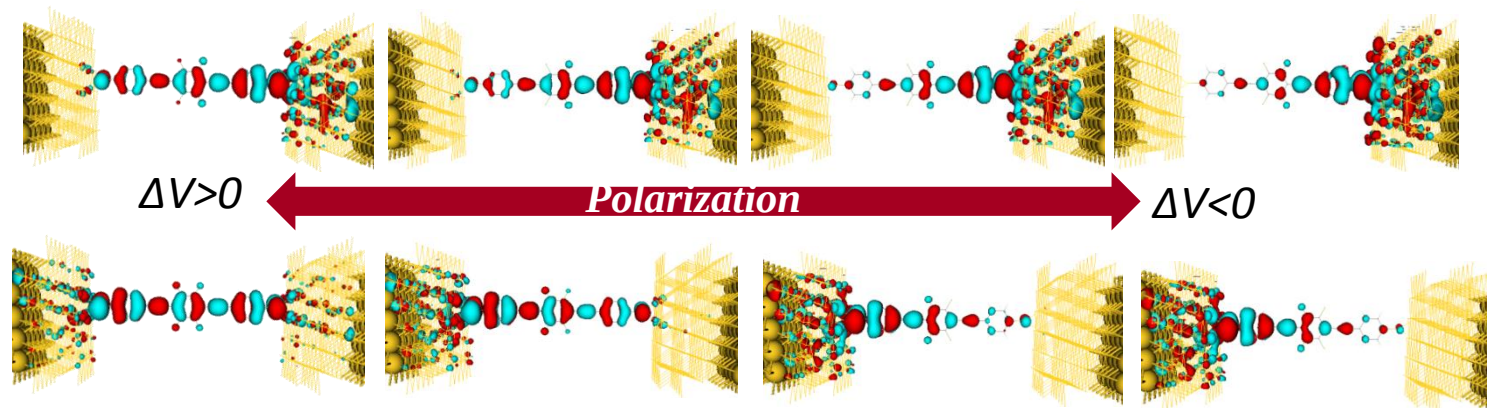


Application of Bias

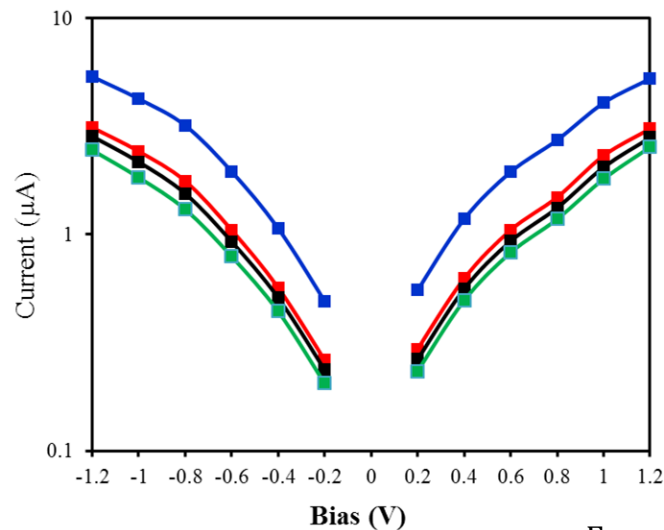
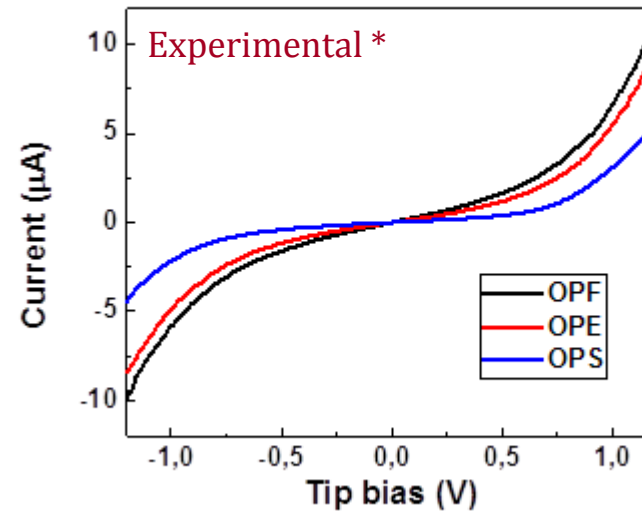
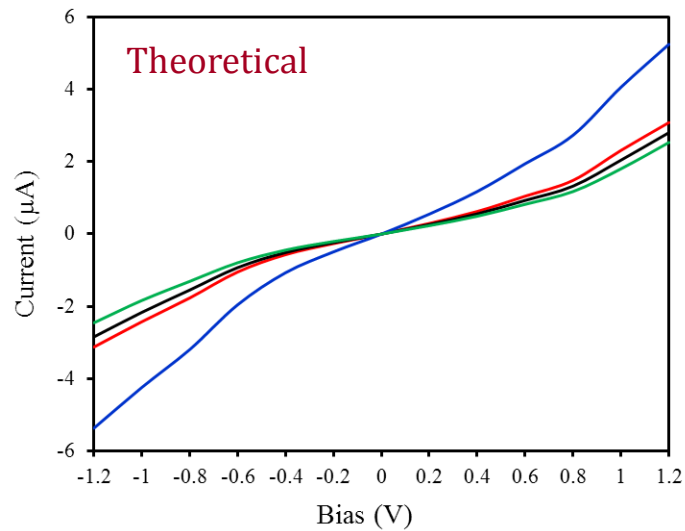


HOMO follows the destabilized electrode! ➡

Intramolecular > Interfacial Coupling



I-V Curves



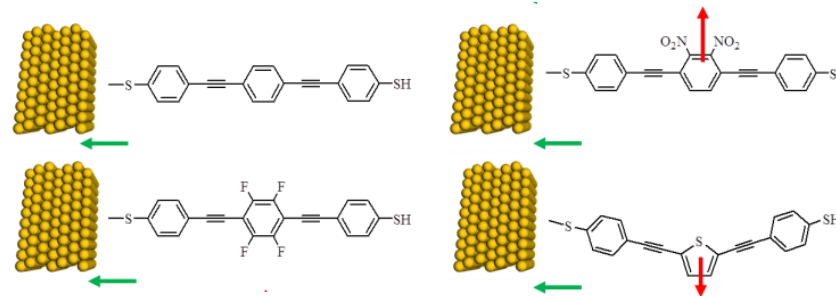
**No impact of the
substituents in lineal
molecules on the I-V curves**



Experimental curves obtained by means CP-AFM Junction, geometry is unknown

Conclusions

1.- Φ modulation is dominated by changes in the net moment dipole along the perpendicular direction to metal surface.



2.- The appearance of interface dipoles, makes the energy of the dominant molecular transmission channel, insensitive to functional group substitutions, pointing a **Fermi Level Pinning Effect**.

3.- The final I/V response is due to a compromise between the degree of orbital polarization/hybridization with the electrodes, governing the screening or amplify the electric field across the molecular backbone.

4.- Theory allows rationalize the I/V characteristics in well-defined molecular architectures, in contrast to many experimental results using different techniques and lacking details about the exact contact geometry or the number of molecules in the junction.

Acknowledgements

Collaborators:

C. Daniel Frisbie

Zuoti Xie

Stéphane Rigaut

Karine Costuas



Silvio Osella

Colin Van Dyck



Northwestern
University

Jéôme Cornil

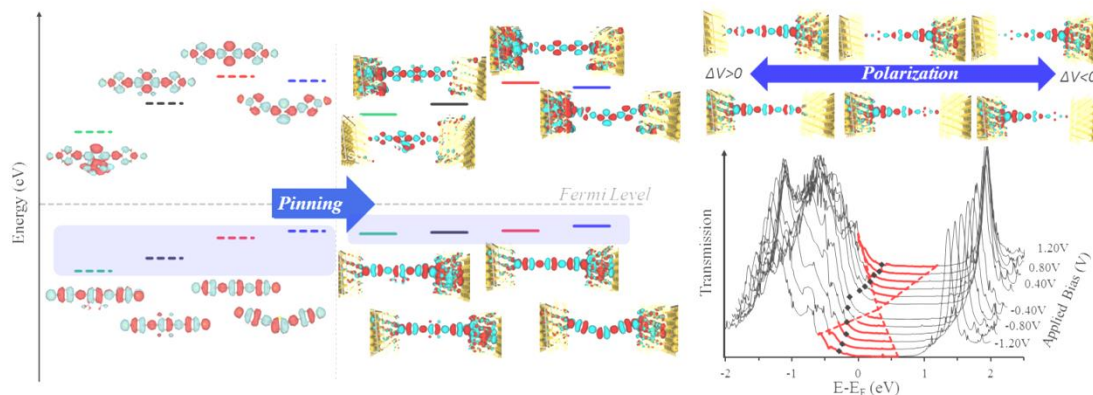
Membres de la bureau B154

UMONS



Organizing committee:





Fermi Level Pinning and Orbital Polarization Effects in Molecular Junctions

Sandra Rodríguez González, Jérôme Cornil, Colin Van Dyck

Sandra.RODRIGUEZGONZALEZ@umons.ac.be

Laboratory for Chemistry of Novel Materials