



Self-assembly and Guided self-organization of oxide nanostructures from chemical solutions

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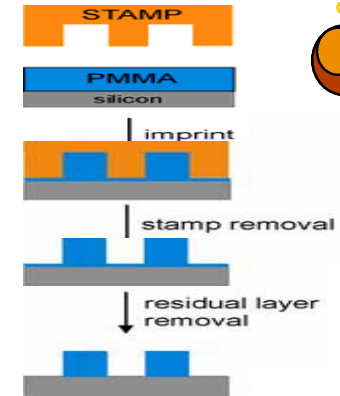
MOTIVATION AND AIM OF THE WORK

- **Nanotechnology** aims the low-cost and mass fabrication of increasingly small devices required for present and future applications

APPROACHES

Top-down

- e-beam lithography
- X-ray/UV lithography
- μ C-printing
- Focused Ion Beam
- Nanoimprint Lithography (NIL)
- SPM...

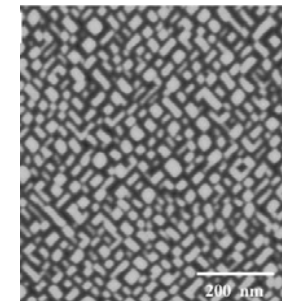


NIL process

Bottom-Up

Self-assembly and Self-organized growth at surfaces

- supramolecular engineering
- metallic, semiconductor and oxide nanostructures.



$\text{Si}_x\text{Ge}_{1-x}$ grown over Si (111) substrate

- Oxide nanostructured templates



exhibit interesting properties for

Our interest: self-assembling of oxide nanostructures with substrate modification through

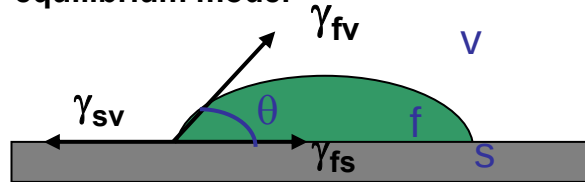
NANOINDENTATION, Focused Ion beam, ...

superconductivity...



Working Framework: THERMODYNAMICS

Liquid nucleus in equilibrium model



Young's equation $\gamma_{sv} = \gamma_{fs} + \gamma_{fv} \cos \theta$



3 Growth Modes

•3D growth (Volmer-Weber)

$$\gamma_{sv} < \gamma_{fs} + \gamma_{fv}$$



•2D (Frank van der Merwe) growth

$$\gamma_{sv} > \gamma_{fs} + \gamma_{fv}$$



•Layer + Island growth (Stranski-Krastanov)

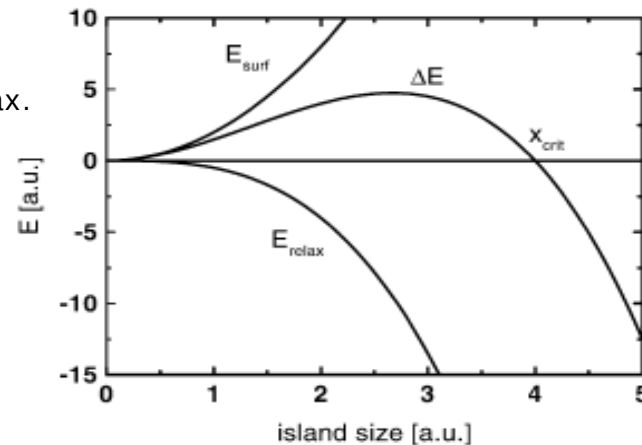
(ex; when strain energy due to lattice mismatch is present)



Heteroepitaxy

$\left\{ \begin{array}{l} \text{Surface Free Energy } (\sim \gamma) \\ \text{Elastic Relaxation Energy } (\sim \varepsilon^2; \varepsilon = (a_s - a_f)/a_f) \end{array} \right.$

$$\Delta E \sim \Delta E_{\text{surf.}} + \Delta E_{\text{elast. relax.}}$$



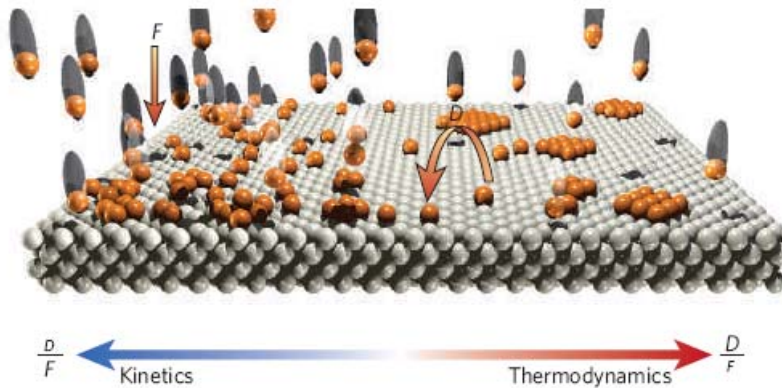
B. Voigtländer/Surface Science Reports 43 (2001)

BUT...



Working Framework: KINETICS

Real growth phenomena are **non-equilibrium**



Barth et al. NATURE 2005 Vol 437 | 29

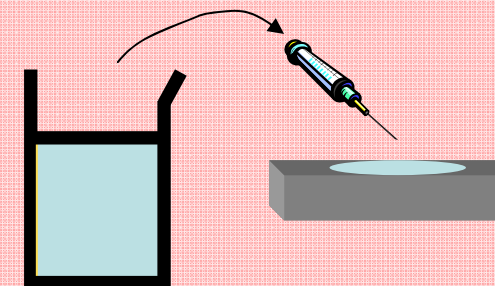
D $\Rightarrow$ Diffusivity

F $\Rightarrow$ Deposition rate

Chemical Solution Deposition (CSD)

CSD Vs Vacuum Techniques:

- ✓ Low Cost
- ✓ Scalable
- ✓ Rapid



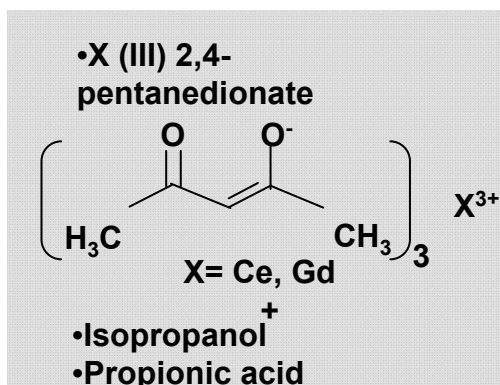
Different initial conditions!



CHEMICAL SOLUTION DEPOSITION (CSD)

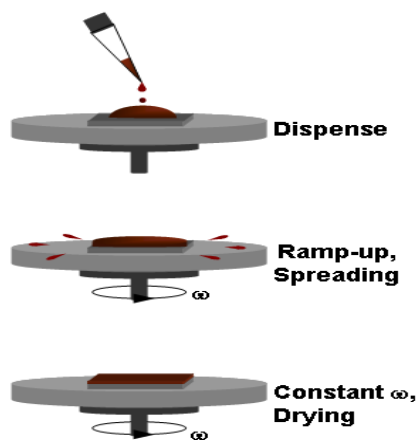
1 Precursor solution synthesis (of a given concentration)

e.g. $\text{Ce}_{1-x}\text{Gd}_x\text{O}_{2-y}$ metallorganic precursor solution preparation (MOD-CSD)



2 Solution Deposition by

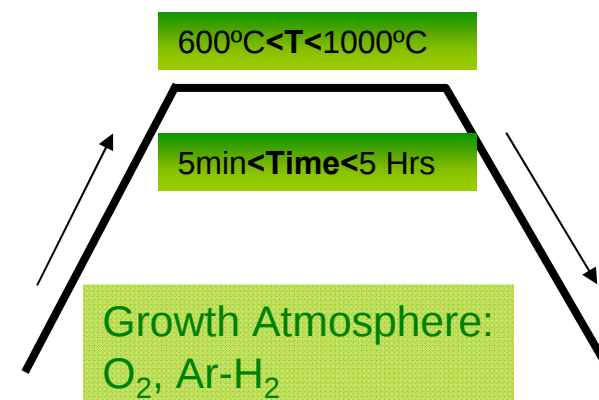
Spin Coating onto the substrate



3 Heat treatment

PYROLISIS of the organic species
(~300°C/400°C)

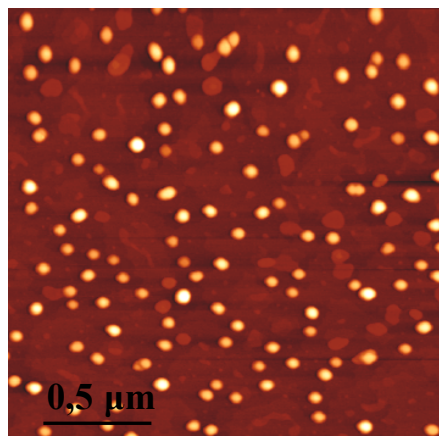
High temperature treatment for the amorphous layer crystallization



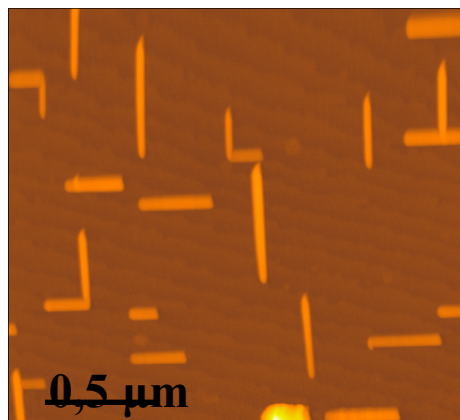
Islands' characteristics (size, morphology,...) can be tuned through the **kinetic** parameters (temperature, atmosphere, dwell time,...)

CSD: A general tool for nanostructured template fabrication

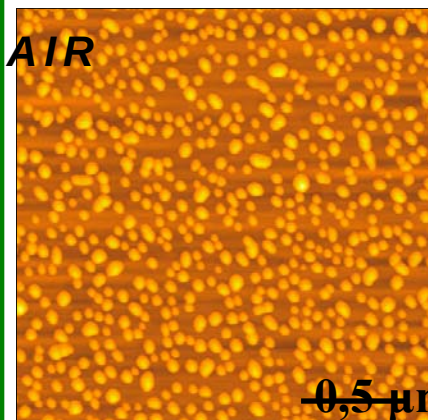
Fluorite: $\text{Ce}_{1-x}\text{Gd}_x\text{O}_{2-y}$



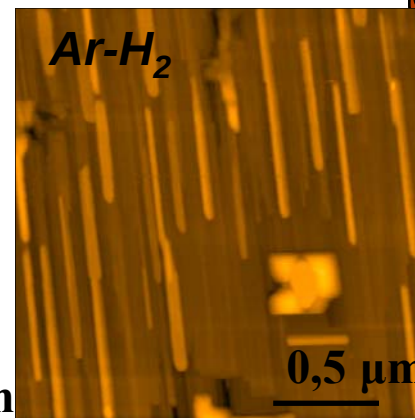
YSZ_{fluorite}



LaAlO₃ perovskite

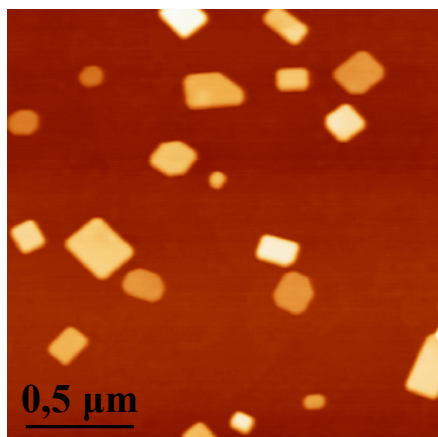


NdGaO₃ perovskite



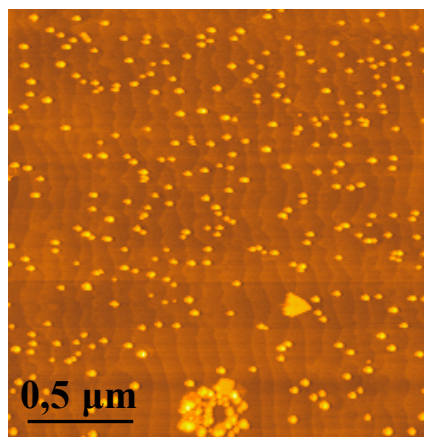
NdGaO₃ perovskite

Cubic: Re_2O_3



LaAlO₃ perovskite

Perovskite: BaZrO_3



SrTiO₃ perovskite

Strain induced
interfacial self-
assembled
nanostructures

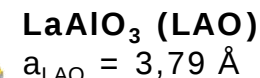
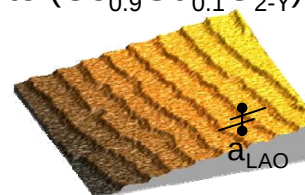
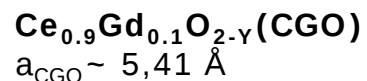
M. Gibert et al.; Surf. Science (2007) 2680
M. Gibert et al.; Adv.Mat. (2007) 3937

Ce_{1-x}Gd_xO_{2-y}(CGO)/LaAlO₃: a system of interest

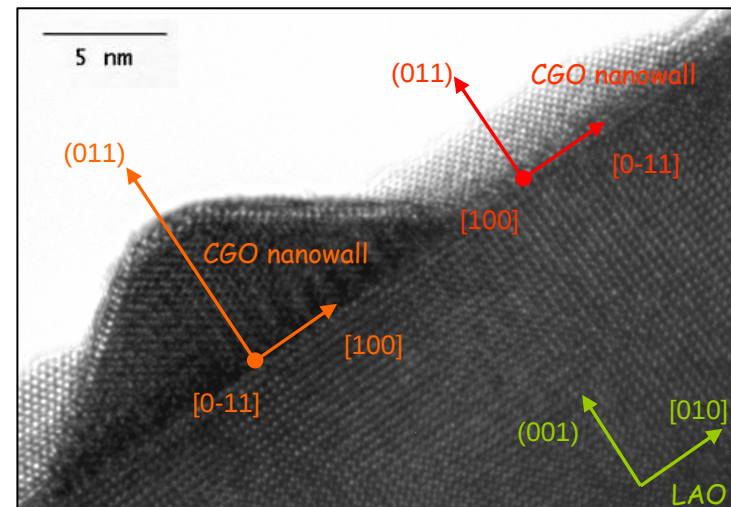
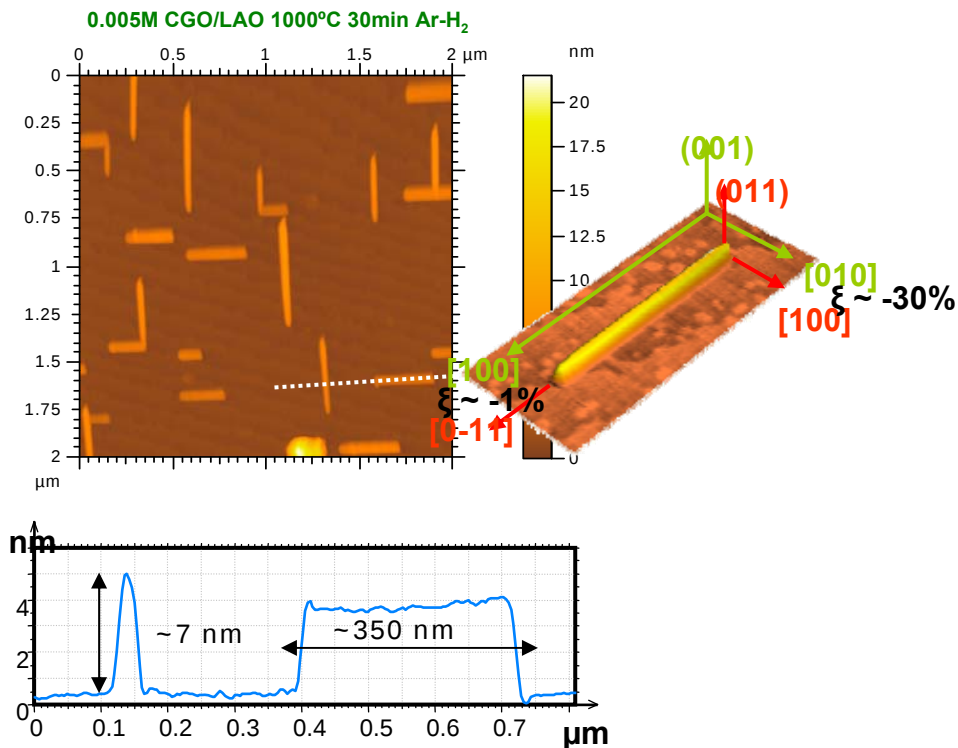
• Wide range of applications:

catalysis, optical properties, ionic conductivity, dielectric properties, buffer layer, nanostructured templates for high T_c superconductors...

• Oxide Heteroepitaxy: fluorite (Ce_{0.9}Gd_{0.1}O_{2-y}) over perovskite (LaAlO₃) structure



Thermally treated perovskite single crystal substrates show atomically flat terraces

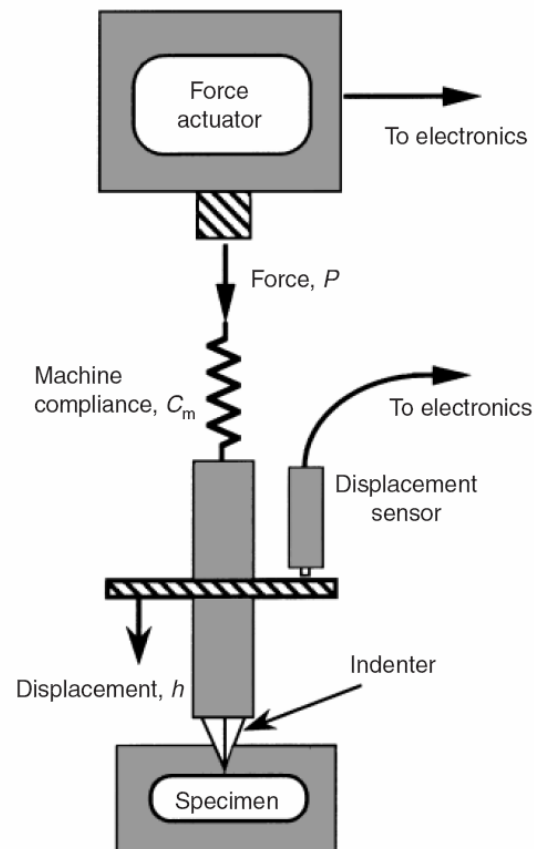


A. Benedetti, F. Sandiumenge
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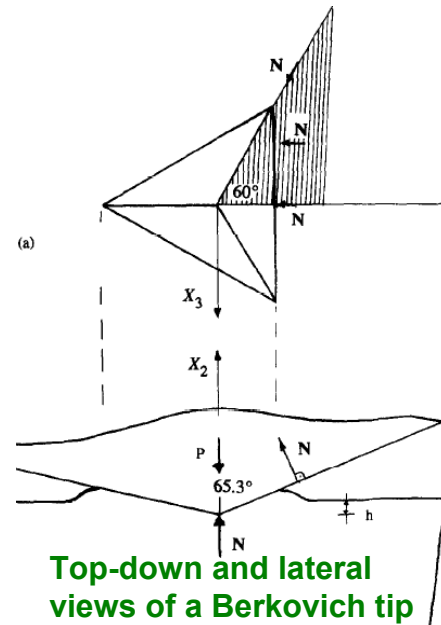
NANOINDENTATION: A tool for nanopatterning

- A sharp diamond tip is driven into and withdrawn from the material
- Ultra low loads are used ($< 10\text{mN}$) and small deformations induced ($< 100\text{nm}$)

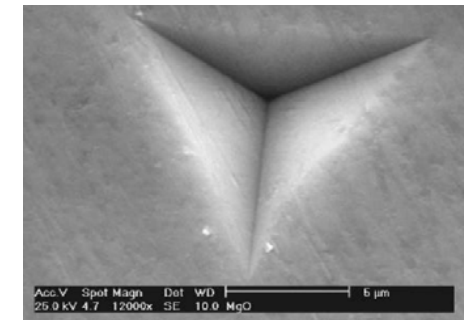
The equipment



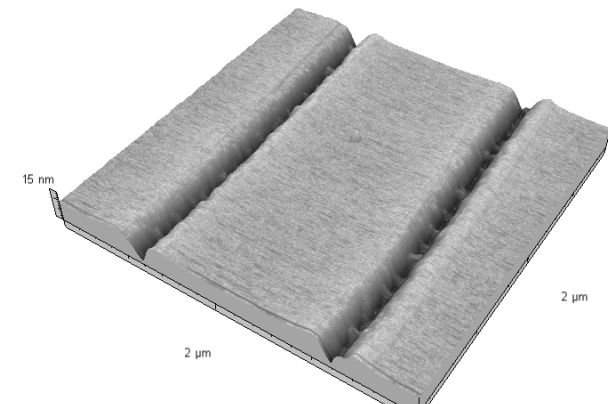
The tip



The Indentation



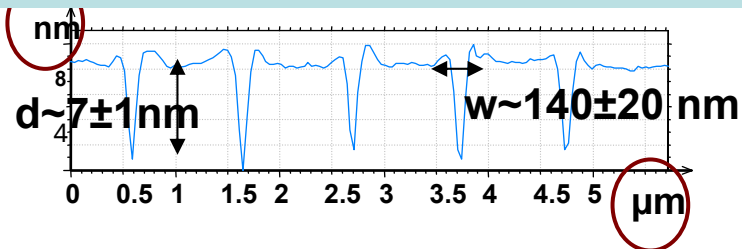
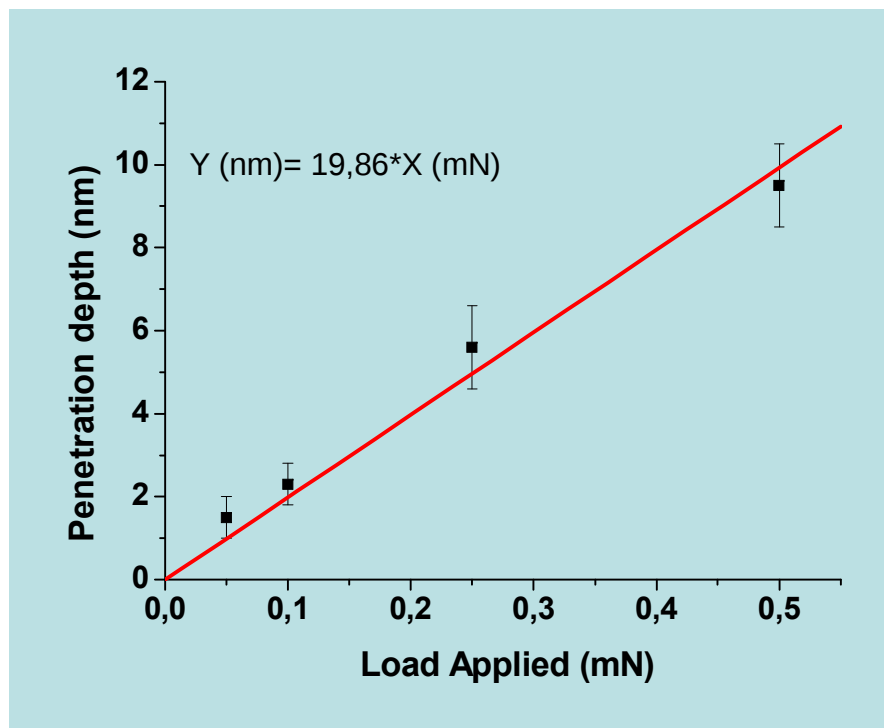
The Scratch



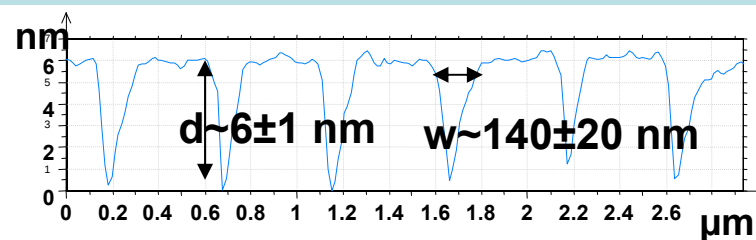
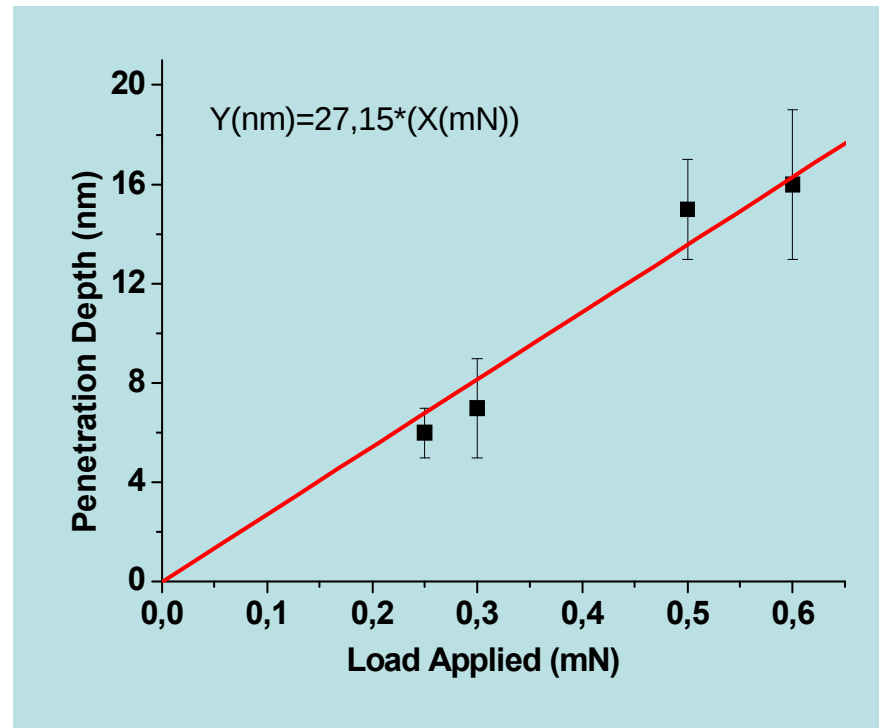


Nanoindentation on perovskite type substrates: tuning of the parameters

SCRATCH



INDENTATION

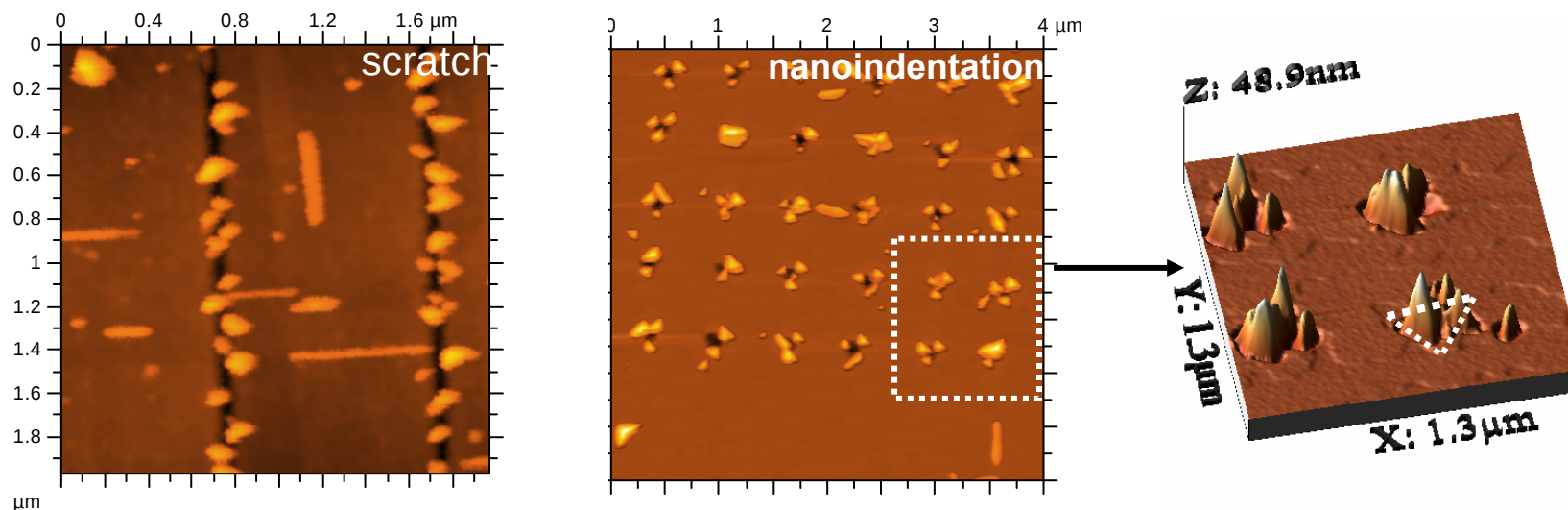


Extremely low-loads in the **0.05mN-0.6mN** range are of interest

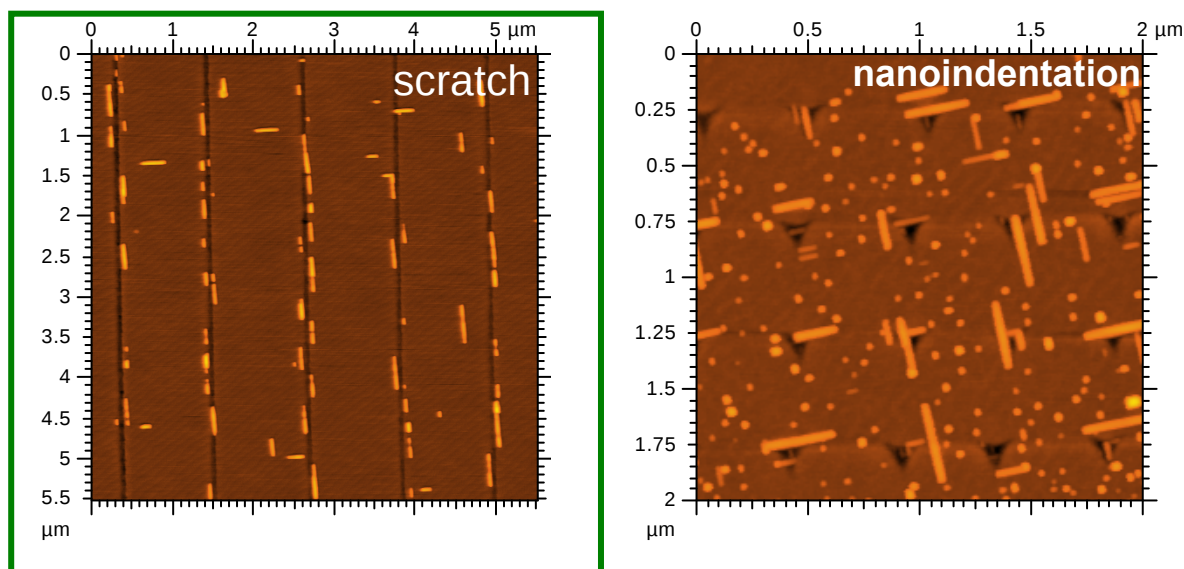


Towards guided self-organization through Nanoindentation

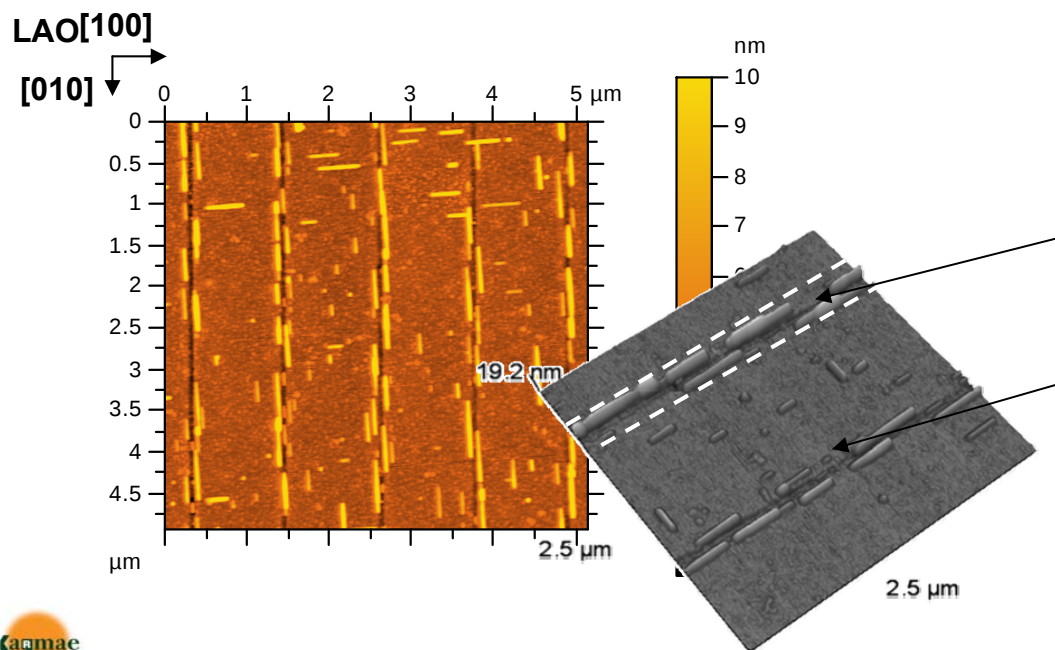
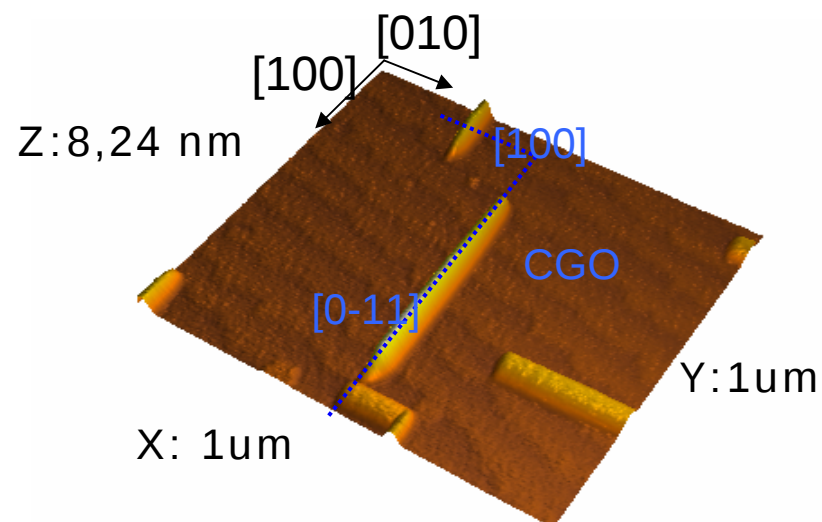
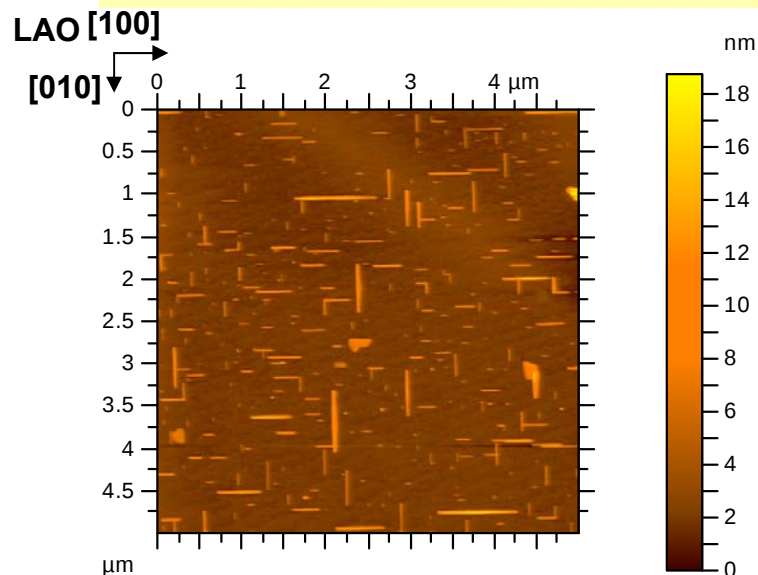
CGO on SrTiO₃ (STO)



CGO on LaAlO₃ (LAO)



CGO nanowalls on LAO: the influence of scratches



0,1 mN Scratches	POPULATION DISTRIBUTION	ISLAND- OCCUPIED AREA
At Indentations	CGO [0-11] //LAO [010] ~ 99%	~40%
In between indentation lines	CGO [0-11] //LAO [010] ~ 60%	~ 5%

Curvature (κ) dependent energy balance

$$E_{\text{island}} \sim E_{\text{surface}} + E_{\text{interface}} + E_{\text{elastic relax}} + \dots$$

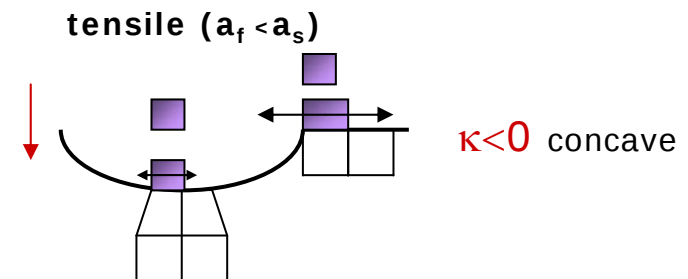
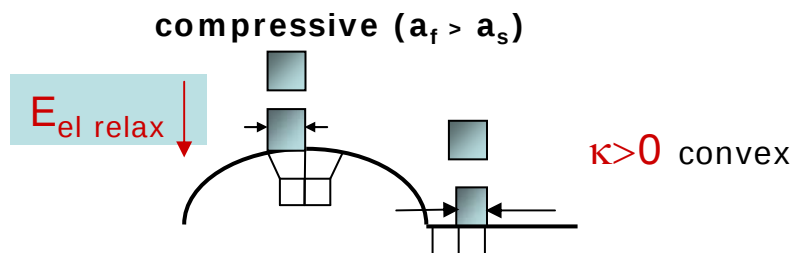
$\sim f(\kappa)$
 $\sim f(\kappa^2, \varepsilon^2)$



E_{int} : related to chemical bonding among the interface atoms



$E_{\text{elast relax}}$: related to the lattice mismatch between the solid crystalline materials

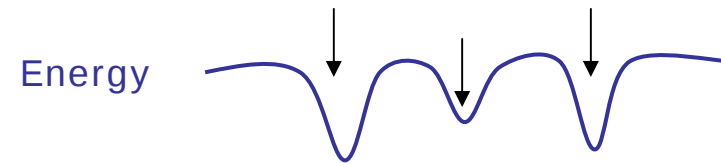
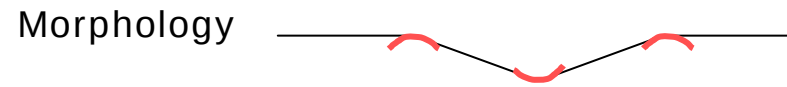
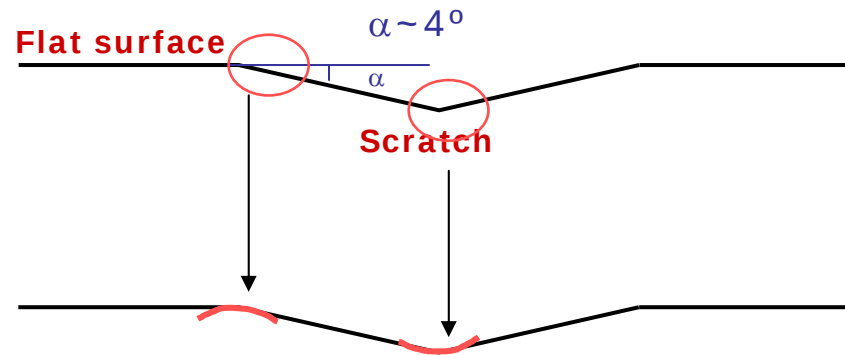
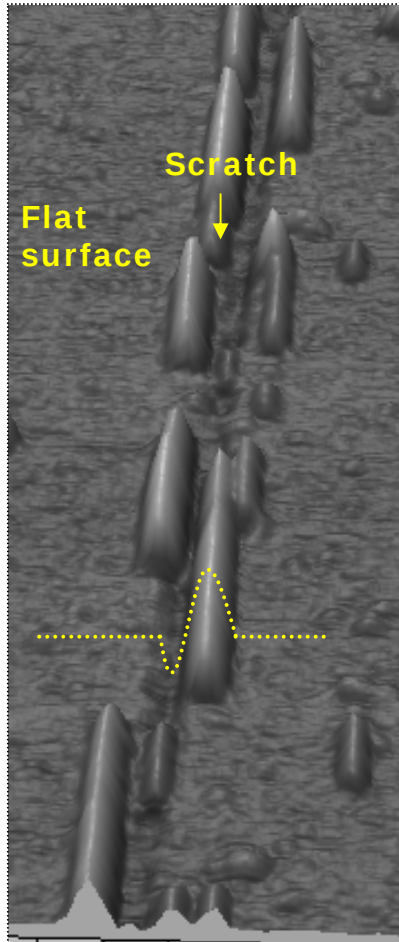


CGO/LAO grows under compressive strain



Concave and **convex** surfaces compete in lowering the local energy with respect to the flat substrate system's energy

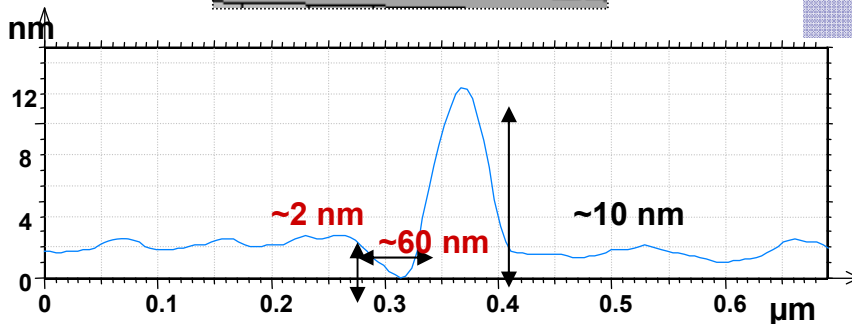
Nanowall LOCATION



Competition between the **surface-energy** and **strain-energy** terms leads to multiple **local minima**



SUBSTRATE MORPHOLOGY INDUCED
DIRECTED ORDERING

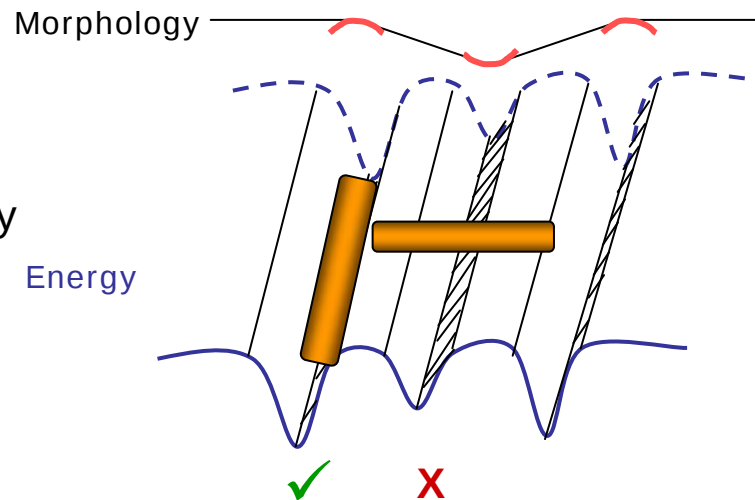




Nanowall ORIENTATION

Why do nanowalls align their long axis with the Scratch direction?

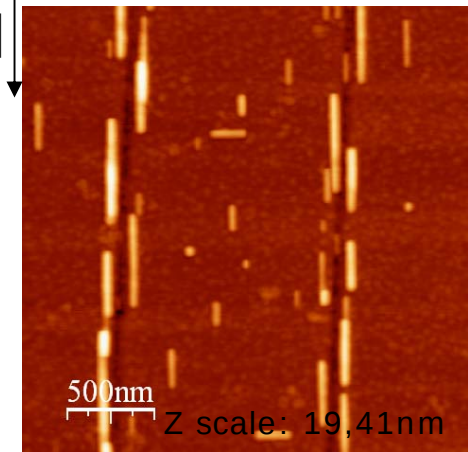
The orientation in which the nanowall maximizes its location at an energy minimum will be favoured



No "horizontal" nanowalls at the scratches

LAO [100]

[010]



CONCLUSIONS

- ✓ **CSD** has been shown a general method for generation of self assembled oxide nanoislands
- ✓ Nanoindentation has successfully been used as a nanopatterning technique
- ✓ **Guided Self-Organized** $\text{Ce}_{0.9}\text{Gd}_{0.1}\text{O}_{2-y}$ nanoisland growth has been evidenced
- ✓ The substrate **curvature effect** on the different energy contributions has been proposed to be responsible for the results observed in the CGO/LAO system
 - nanowall **location** at scratch edges is evidenced
 - pre-existing nanowall **orientation** balance is broken

The nanoislands' kinetic evolution, under study, seems to give very rich phenomena suggesting further influence of the nanoindentation parameters

THANK YOU!