

Surface anisotropy, orbital moment and biomedical applications in magnetic nanoparticles

N. Pérez^(a), P. Guardia^(a), A.G. Roca^(b), M.P. Morales^(b), C.J. Serna^(b), F. Bartolomé^(c), L.M. García^(c), J. Bartolomé^(c), A. Labarta^(a), X. Batlle^(a)

^a *Dept. Física Fonamental and Insitut de Nanociència i Nanotecnologia IN2UB, U. Barcelona, Martí i Franqués 1, 08028 Barcelona, Spain*

^b *ICMM-CSIC, Sor Juana Inés de la Cruz 3, Cantoblanco 28049, Madrid, Spain*

^c *Dept. Física de la Materia Condensada, U. Zaragoza and ICMA-CSIC, Pedro Cerbuna 12, 50009 Zaragoza, Spain*

nicolas@ffn.ub.es

Magnetic nanoparticles (NP) systems have long been subject to study finite-size and surface effects [1]. Besides, their potential application for biomedical purposes relies on high quality magnetic materials. The thermal decomposition of an organic iron precursor in an organic medium [2] allows the preparation of highly crystalline iron oxide NP with excellent magnetic parameters [3]. Particles in the 5-50 nm range were synthesized in the presence of a variety of coatings with controlled shapes. All the materials show a narrow size distribution with high crystal quality. Saturation magnetization was size independent in the 5-20 nm range and almost reached the expected value for bulk magnetite at low temperatures, higher in those NP with the surfactant covalently bonded to the surface. In 5 nm particles the surface contribution to magnetic anisotropy could be established via an analytical method that relies on the $T \ln(t/t_0)$ scaling and demonstrates that surface anisotropy causes the broadening of their energy barrier distribution [4,5]. X-ray absorption spectra (XAS) suggested charge transfer from the NP to the covalent bonded surfactant. X-ray magnetic circular dichroism (XMCD) confirmed the dependence of the magnetic moment on the surface bond and suggested that the orbital momentum is more effectively quenched in covalently bonded NPs. Besides, the low-temperature $\langle S_z \rangle = 3.63 \mu_B/\text{f.u.}$ obtained in the latter, is very close to those reported for bulk samples (3.90-3.95 $\mu_B/\text{f.u.}$). High resolution TEM suggests that the foregoing is related to the crystal quality of the NP [6].

Acknowledgements

MEC (NAN2004- 08805-CO4-02, NAN2004-08805-CO4-01, CONSOLIDER CSD2006-12, MAT2005-02454 and MAT2006-03999), Generalitat de Catalunya (2005SGR00969)

References:

- [1] X. Batlle and A. Labarta, J. Phys D: Appl. Phys 35, R15 (2002)
- [2] J. Park et al., Nat. Mat.3, 891 (2004)
- [3] P. Guardia et al., J. Magn. Magn. Mat. 316, e756 (2007)
- [4] N. Pérez et al., Nanotechnology 19, 475704 (2008)
- [5] O. Iglesias et al., J. Magn. Magn. Mat. 140-144, 399 (1995)
- [6] N. Pérez et al., Appl. Phys. Lett. 94, 093108 (2008)